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PATENT KANTOOR DEPARTEMENT VAN HANDEL EN NYWERHEID REPUBLIC OF SOUTH AFRICA

PATENT OFFICE DEPARTMENT OF TRADE AND INDUSTRY

Hiermee word gesertifiseer dat This is to certify that

- 1 South African Provisional Patent Application No. 2003/8306 accompanied by a Provisional Specification was originally filed at the South African Patent Office on 24 October 2003, in the name of MISTY MOUNTAIN TRADING 19 (PTY) LIMITED in respect of an invention entitled: WARE-WASHING PROCESS.
- On 26 November 2003 a change of name from MISTY MOUTAIN TRADING 19 (PTY) LIMITED to OZONE WASH (PTY) LIMITED was recorded at the Companies Office.
- 3. The photocopy attached hereto is a true copy of the provisional specification and drawings filed with South African Patent Application No. 2003/8306.

Geteken te

Signed at

in die Republiek van Suid-Afrika, hierdie

**PRETORIA** 

in the Republic of South Africa, this

dag van

December 2004

day of

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Registrar of Patents

PRIORITY DOCUMENT

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REGISTRATEUR VAN PATENTE, MODELLE, HANDELSMERKE EN OUTEURSREG

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#### **REPUBLIC OF SOUTH AFRICA**

PATENTS ACT, 1978

# PROVISIONAL SPECIFICATION

(Section 30 (1) - Regulation 27)

OFFICIAL APPLICATION NO.	LODGING DATE	DMK REFERENCE
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	FULL NAME(S) OF APPLICANT(S)  MISTY MOUNTAIN TRADING 19 (PTY) LI  FULL NAME(S) OF INVENTOR(S)  FLETCHER, Clive Robert  ACKAY, Donald Alexandrian  TITLE OF INVENTION  WARE-WASHING PROCESS	FULL NAME(S) OF APPLICANT(S)  MISTY MOUNTAIN TRADING 19 (PTY) LIMITED  FULL NAME(S) OF INVENTOR(S)  FLETCHER, Clive Robert  ACKAY, Donald Alexander  TITLE OF INVENTION  WARE-WASHING PROCESS

## **WARE-WASHING PROCESS**

#### Field of the Invention

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This invention relates to a ware-washing process. More particularly, this invention relates to a ware-washing process that is conducted at low wash water temperatures.

# **Background to the Invention**

In this specification, the use of the word "ware" is taken to mean items such as wares, cutlery and crockery, pots and pans, and associated items of tableware and eating utensils.

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The success of conventional ware-washing processes and the associated use of conventional detergents tends to be found in their removal of common food soils under alkali conditions, using inorganic alkali. While these processes remove a large number of fats, proteins and sugars, due to the solubility of such soils in water, it is well documented that these soils lend themselves, primarily, to removal under warm or even hot conditions. Industrial auto-warewashing is presently, conventionally a warm water multi-stage process, including a prewash stage (typically conducted at temperatures of 30°C – 40°C), a detergent wash stage (typically conducted at temperatures of 55°C – 65°C) and a rinse stage (typically conducted at temperatures of approximately 85°C). The current practice requires operation of washing machines at these relatively high temperatures in order to ensure, first, the breaking of chemical and/or physical bonds between the soils and the item to be washed and, second, the precipitating out of solution of those soils via a chelation or sequestration process.

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In this specification, the term "warm water washing" is understood to mean washing at temperatures typically in the range  $30^{\circ}$ C –  $85^{\circ}$ C, while the analogous term "cold water washing" is understood to mean washing at temperatures typically in the range  $10^{\circ}$ C –  $25^{\circ}$ C.

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Use of conventional detergents in warm water washing has several well-documented disadvantages. For example: the rate of foaming of a chemical detergent, typically, is proportional to the wash water temperature employed. Accordingly, higher wash water temperatures result in greater foaming, which is an undesired result mechanically, for example, in the warewashing industry.

A further disadvantage of conventional warm water washing detergents is the increased energy-consumption associated with generating hot water, as well as the increased down-time and maintenance of washing machine components, including boilers and elements, that is required, relative to cold-wash machines.

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Another disadvantage associated with warm water washing processes is the fact that they tend to generate relatively hot to humid conditions in the machine, which conditions are conducive to the sustainability of various forms of bio organisms such as bacteria, algae, fungi and moulds. These same conditions also create a habitable environment that is favoured by pests such as cockroaches. Accordingly, warm water warewashing processes tend to lend themselves to at least some objectionable, unhygienic consequences.

The use of biocides in a warm water washing process is known. It is known that, while ozone is an effective biocide against bacteria and algae, ozone also has several inherent difficulties, including the fact that it is highly reactive, and that it is difficult to localise, since it is a gas at ambient temperatures. For these reasons, ozone has yet to be considered for successful use or used as a biocide in the warewashing industry.

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#### Object of the Invention

It is an object of the present invention to provide a novel warewashing process that, it is believed, will overcome or at least minimize the disadvantages and difficulties associated with prior art as set out above.

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#### Summary of the Invention

According to a first aspect of the invention, there is provided a cold water warewashing process including the steps of:

- (i) washing the items in the warewasher in a cold water wash cycle; and
- (ii) rinsing the items in the warewasher in a rinse cycle;

the process being characterised by the inclusion of the step of introducing a biocide into the washing cavity of the warewasher.

The biocide may be introduced in a gaseous phase.

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The biocide is preferably ozone.

The biocide may be introduced at a temperature in the range 15°C - 25°C.

The biocide may be introduced at a pH range between 2 - 12.

5 The cold water warewashing process may include the step of pre-rinsing the items in the warewasher.

The cold water warewashing process may include the step of recycling the biocideinfused water into the washing cavity.

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The cold water wash cycle may involve the step of using a warewashing detergent composition comprising:

- an inorganic caustic alkali; and
- a complexing agent,
- the inorganic caustic alkali and the complexing agent combining to clean wares in a cold water washing process.

The warewashing detergent composition may be provided in the ratio of between 0.1% – 55% inorganic caustic alkali, between 0.1% – 45% complexing agent.

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The inorganic caustic alkali component of the warewashing detergent composition may be a compound of the class of compounds selected from the group consisting of: alkali earth metal hydroxides, and any combination thereof.

- The complexing agent may be a compound of the class of compounds selected from the group consisting of: complex phosphates, simple phosphates, amino-carboxylic acids, nitrolo triacetic acid (NTA), phosphonic acids, phosphonobutones, acrylates, and any combination thereof. Preferably, the complexing agent is NTA.
- Preferably, the inorganic caustic alkali component of the warewashing detergent composition is sodium hydroxide, alternatively, potassium hydroxide.

The rinse cycle may involve the use of a rinse aid composition, preferably comprising:

- 35 an alkoxylate; and
  - an acid,

the alkoxylate and acid combining to rinse wares at low wash water temperatures and without streaking.

The low-temperature rinse aid composition may be provided in the ratio of between 0.1% – 90% alkoxylate, and between 0.1% – 25% acid.

The alkoxylate component of the rinse aid composition may be an alcohol alkoxylate.

The chain length of the alcohol alkoxylate component of the rinse aid composition may be varied between C<sub>4</sub> - C<sub>22</sub>.

The degree of ethoxylation of the alcohol alkoxylate component of the rinse aid composition may be varied between 1 mole to 30 mole ethylene oxide. Preferably, the alkoxylate component of the rinse aid composition is an alcohol alkoxylate being 100% active with the cloud point of a 1% solution of temperature less than 22°C.

The acid component of the rinse aid composition may be selected from the group of acids consisting of: citric acid, acetic acid, sulfamic acid, phosphoric acid, and any combination thereof. Preferably, the acid component of the rinse aid composition is citric acid.

The Rinse Aid composition may include a dye.

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According to a second aspect of the invention, there is provided the use of ozone in a cold water warewashing process. The cold water warewashing process preferably is as hereinbefore defined.

According to a third aspect of the invention there is provided a warewasher having a washing cavity, the warewasher including introduction means for introducing ozone into the washing cavity.

## **Detailed Description of the Invention**

Without limiting the scope of the invention and by means of examples only, embodiments of the invention will now be described hereunder.

5 <u>Example 1: Experimental Procedure for the Determination of the Efficacy of Ozone</u> as a Biocide in Ware Washing

Each Formulation was evaluated using a Hobart F25 Warewashing machine.

More particularly, the Hobart F25 warewashing machine used had been modified by disconnecting the heating elements so that the warewashing process could only be carried out at ambient temperatures within the cold water washing range. The particular results discussed below were conducted and recorded at an ambient temperature of 17.4°C.

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The Hobart F25 Warewashing Machine operated on a 6 minute warewashing process that included a pre-rinse cycle, a wash cycle and a rinse cycle. The detergent and rinse aid were both dosed automatically into the washing machine.

20 The solutions used were made up to the following concentrations:

- detergent: 50 ml in 50 l of water; and
- rinse aid: 5 ml in 20 l of water.

Uniform standard white dinner plates were used for the experiment. The rate of addition of ozone was 780 mg/hour giving less than 1 ppm concentration and the process of addition of the ozone into the washing machine during the 6 minute wash cycle was conducted at a rate of 780 mg per hour.

Bacteria counts were conducted before and after washing, the results of which are tabulated below in Tables 1 & 2. The trial achieved a kill rate of 100% of all bacteria detected.

#### Ozone Usage Trial in Warewashing Machines

Scope of trial

35 To determine the kill efficacy of ozone on selected bacteria in warewashing machine.

#### Bacteria used

- staphylococcus aureus;
- escherichia coli;.
- pseudomonas auruginosa;.
- bacillus subtilis;
- 5 salmonella typhi; and
  - listeria monocytogenes,

the above bacteria having been sourced from the following batches obtained from the South African Bureau of Standards.

- 10 S. aureus (STA 53)
  - E.coli (SABS TCC ESC 37)
  - P.Aeruginose (PSE 16)
  - B.subtillis (BAC 35)
  - Salmonella (SAL 10)
- 15 Listeria (LI 5)

#### Testing Methodology and Tabulated Results

- The surface of the plate used to conduct the testing on was swabbed prior
   testing, and the plate washed with the cleansing aid known in the trade as T & C Chemical product 96210.
  - 2. The surface of each plate on which testing was to be conducted was divided into two columns with 12 rows each.
- The column on the left hand side was marked "Before Washing" and the
   column on the right hand side was marked "After Washing".
  - 4. Loopfulls of each bacteria culture were placed and suspended into 5 ml separate aliquots of sterile milk, with each such aliquot being tested for the presence of antibiotics. Only those aliquots indicating a negative result for the presence of antibiotics were used in further experimentation.
- 30 5. Each row on each plate was inoculated with solutions of the respective bacteria-types specified above.
  - 6. The plates were left to dry in an incubator for 10-15 minutes.
  - 7. Each row in the column on the left hand side of the plate marked "Before Washing" was swabbed.

**Table 1: Measured Bacteria Counts Before Washing** 

Swab	Total Aerobic	Staph SGS	E.coli	Pseudomon as	Bacillus	Salmonella	Listerla
Description	Count SABS	1TP:012	SGS	spp	Cereus	spp.	OXOID
	763	Count/Area	1TP:004	OXOID 6 <sup>th</sup> EDD	SGS	1TO:018	6 <sup>™</sup> Ed
	Count/Area		Count/Area	(1990)	1TP:011	Count/Area	(1990)
	-		•	Count/Area	Count/Area		Count/Area
Swab 1	0		<del></del>				
Swab 2	0						
Swab 3		>3000					
Swab 4			214				
Swab 5				Present			
Swab 6					Present		
Swab 7						Present	
Swab 8							0
Swab 9							
Swab 10							
Swab 11						· ·· · · · · · · · · · · · · · · · · ·	
Swab 12							
Swab 13							
Swab_14					<del></del>		

- 8. Plates were then washed in the warewashing machine in the presence of ozone, together with the compounds known conventionally in the industry as T & C Chemicals detergent products automatic warewashing aid (No. 92696) and T & C Chemicals automatic rinse aid (No. 92695), and left to dry.
- 9. After the wash cycle was completed, the plates was removed from the warewashing machine and the right hand side of each row in the "After Washing" column was swabbed.

**Table 2: Measured Bacteria Counts After Washing** 

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Swab	Total Aerobic	Staph SGS	E.coli	Pseudomon as	Bacillus	Salmoneila	Listeria
Description	Count SABS	1TP:012	SGS	spp	Cereus	spp.	OXOID
	763	Count/Area	1TP:004	OXOID 6th EDD	SGS	1TO:018	6 <sup>™</sup> Ed
1	Count/Area		Count/Area	(1990)	1TP:011	Count/Area	(1990)
				Count/Area	Count/Area		Count/Area
Swab, 1							
Swab,2							
Swab 3						<del></del>	
Swab 4						-	
Swab 5					<u> </u>		
Swab 6							
Swab 7							

Swab 8			l			T
Swab 9	None	<u>-</u>				<del>                                     </del>
	Detected					
Swab 10		None				<del> </del>
		Detected				1
Swab 11			None Detected			
Swab 12				None		<del></del>
				Detected		· ·
Swab 13					Absent	<del>                                     </del>
Swab 14					<del> </del>	<del>  0</del>

Kill Rate Achieved during this Trial: 100%

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# Example 2: Description of the Cycles in a Preferred Cold Warewashing Process

The warewashing process used in the evaluation of the efficacy of ozone as a biocide in warewashers was a 6 minute cycle which included a pre-rinse cycle, wash cycle and rinse cycle. Both the detergent as well as the rinse aid (as described above) were automatically dosed into the machine at a rate of 50 ml per 20 I water for the detergent and 25 ml per 20 I water for rinse aid. The ozone was dosed into the machine in a vapour phase at a rate of 780 mg per hour. The ozone generating unit was designed to commence dispensing ozone into the washing machine on activation of the washing machine, and to cease operation when the washing cycle had been completed.

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# Example 3: Relative Performance Evaluation: Variable Rates of Introduction of Ozone into the Washing Machine

The experimental procedure as described in Example 1 were repeated, varying only the rate of addition of ozone into the washing machine from 500 mg per hour to 900 mg per hour. Consideration and comparison of the results obtained revealed that the optimum was found at 780 mg per hour.

It is worth noting that the bacterial load used in the experimental tests conducted were exceedingly high and such loads would not normally be found in practice and therefore the ozone value could be reduced significantly.

The optimum dosages for the detergent and rinse aid were found, similarly, to be 50 ml per 20 l water and 25 ml per 25 l water respectively, although this could vary substantially depending on the degree of soiling and also bacterial load.

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It will be appreciated that numerous embodiments of the invention may be performed without departing from the scope and spirit of the invention as defined in the consistory statements above.

Dated this 24 day of Cliffor Joe3

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Patent Attorney Agent for the Applicant